STUDY AND EVALUATION OF ANTIMICROBIAL ACTIVITY AND ANTIOXIDANT CAPACITY OF DRY EXTRACT AND FRACTIONS OF LEAVES OF *Raphanus sativus* var. *oleiferus* Metzg.

ESTUDO QUÍMICO E AVALIAÇÃO DA ATIVIDADE ANTIMICROBIANA E DA CAPACIDADE ANTIOXIDANTE DO EXTRATO SECO E FRAÇÕES DE FOLHAS DE Raphanus sativus *var*. oleiferus *Metzg*.

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ABSTRACT: The radish (Raphanus sativus L.) is a vegetable of the Brassicaceae family cultivated worldwide and has several medicinal properties. Its biological activities are related to various secondary metabolites present in the species, especially phenolics. Thus, the objectives of this study were the chemical analysis and evaluation of the antioxidant and antimicrobial activities of the dry extract and fractions of the fodder turnip leaves (R. sativus var. oleiferus Metzg.). Samples were analyzed by mass spectrometry and the antioxidant activity was evaluated using the 2,2-diphenyl-1-picrylhydrazyl (DPPH) radical method and the reducing power method. Antimicrobial activity was determined by the agar diffusion and microdilution methods. The total phenols were concentrated in the butanol fraction (121.27 mg GAE/g) and the flavonoids were concentrated in the ethyl acetate fraction (98.02 mg EQ/g). The ethyl acetate fraction showed the best antioxidants results, with 83.45% of free radical scavenging and 11.34% of ferric ions reduction. The analysis of antimicrobial activity showed that the dry extract had the highest average zone of inhibition against Bacillus subtilis (18.67 mm). Smaller values of the minimum inhibitory concentration for Micrococcus luteus were, and the ethyl acetate fraction showed a lower minimum inhibitory concentration (0.1 mg/ml) for that microorganism. There was a strong correlation between the antioxidant activity and the content of phenols and flavonoids. The results showed the potential antioxidant and antimicrobial activities of this extract with the ethyl acetate fraction being most promising for further studies.

KEYWORDS: Fodder turnip. Phenolic compounds. Brassicaceae. Medicinal plants.

INTRODUCTION

The use of plants for medicinal purposes is an old practice and is associated with popular knowledge from different parts of the world (VEIGA; PINTO; MACIEL, 2005). The study of the chemical composition of plants indicates their potential biological properties and can encompass both the therapeutic and toxicological properties (SOUZA-MOREIRA; SALGADO; PIETRO, 2010).

The discovery of new plant products can significantly contribute to global health, as well as providing alternative therapeutic approaches. There is also the possibility of isolating substances with greater efficacy, lower cost, lower toxicity, milder side effects and greater availability of the raw material (MENEZES et al., 2009).

The Brassicaceae family has about 350 genera and 4000 species, originating in the northern

hemisphere and distributed in temperate zones around the New World (ZECCA, 2008). Several horticultural varieties are found, especially *Brassica oleracea* var. *acephala* (kale), *B. oleracea* var. *capitata* (cabbage), *B. oleracea* var. *borytis* (cauliflower), *B. oleracea* var. *italica* (broccoli), *B. nigra* and *Sinapis* sp. (mustard), *Raphanus sativus* L. *var. radicle* (radish) and *Eruca sativa* (arugula) (SOUZA; LORENZI, 2005).

The common radish (R. sativus L.) is a vegetable root crop with numerous varieties, among them: radicle. niger, mougri and oleifera (GUTIERREZ; PEREZ. 2004; ARUNA: YERRAGUNI; RAJU, 2012). Different parts of the plant (leaves, seeds and roots) are used in folk medicine for gastrointestinal disorders and biliary, liver, urinary, respiratory and cardiovascular ailments (DEVARAJ; KRISHNA; VISWANATHA, 2011; SHIN et al., 2015). In addition, the plant

606

contains bioactives with anti-inflammatory (PARK; SONG, 2017; MANIVANNAN et al., 2019), antimicrobiana, antioxidante, antimutagênica (GUTIERREZ and PEREZ, 2004; ALQASOUMI; AI-HOWIRINY; RAFATULLAH, 2008; SHIN et al., 2015; LUO et al., 2018; MANIVANNAN et al., 2019), antiviral (GUTIERREZ; PEREZ, 2004; SHIN et al, 2015; PARK et al., 2017) e antidiabética (BANIHANI, 2017; MANIVANNAN et al., 2019) properties. The major secondary metabolites of the species are polyphenols, flavonoids, alkaloids, tannins, volatile oils and glycosinolates (ARUNA; YERRAGUNI; RAJU. 2012). However the flavonoids are undoubtedly are the largest group of bioactive chemicals (JAHANGIR et al., 2009).

R. sativus var. *oleiferus* Metzg., popularly known as fodder turnip, is a variety of radish little studied from the chemical and biological point of view. Therefore, this study aimed to perform chemical analysis and to evaluate the antioxidant and antimicrobial potential of fodder turnip, in order to correlate the main chemical constituents detected with the biological activities.

MATERIAL AND METHODS

Plant material

The leaves of fodder turnip were collected in June 2013, on a plot situated at Avenida Afonso Pena, the city of Alfenas (latitude 21°25'22.87"S, longitude 45°57'29.42"W), Minas Gerais, Brazil. The climate of the site at the time was cold and dry, with an average temperature of 14 °C. The identity of the plant was confirmed by Prof. Dr. Geraldo Alves da Silva. A voucher specimen of the plant material was deposited in the Herbarium of the Federal University of Alfenas and registered under number 2279.

Preparation of the dry extract and fractions

The leaves were dried in an oven with air circulation at 45 °C for 72 hours and then pulverized. Three hundred grams of plant material was submitted to the percolation method (PRISTA; ALVES; MORGADO, 1992) using a mixture of ethanol: water (7:3, v/v) as liquid extactor. Subsequently, the extract obtained was concentrated on a rotaevaporator and finally dried by lyophilization, with a final yield of 29,198%. Ten grams of the dry extract was solubilized in water and subjected to a process of liquid-liquid extraction with increasingly polar solvents. After extraction and removal of the organic solvent the hexane (FrHex), ethyl acetate (FrAcOEt), butanol (FrBuOH) and aqueous (FrAq) fractions were

obtained. The yields from each fraction compared to the initial amount of dry extract were 10.26%, 5.88%, 30.40% and 22.12%, respectively.

Analysis by mass spectrometry with electrospray ionization (ESI-MS)

Mass spectra were obtained using an LTQ XL Linear Thermo Scientific 2D mass spectrometer, equipped with direct insertion of the sample via continuous flow injection (FIA). The extract and the fractions were analyzed in electrospray ionization mode (ESI) with fragmentation in multiple stages held in an ion trap type interface (IT). Negative mode was selected for the generation and analysis of the first-order mass spectra (MS) as well as for other experiments in multiple stages (MSⁿ) using the following conditions: capillary voltage of -41 V voltage spray -5 kV, capillary temperature 280 °C, carrier gas (N_2) flow 60 (arbitrary units). The track acquisition was m/z 50-2000 with two or more sweep events performed simultaneously in the LTQ XL mass spectrometer.

Determination of total phenols

The total phenolic content was determined by the colorimetric method of Folin-Ciocalteau (SINGLETON; ORTHOFER; LAMUELA-RAVENTOS, 1999). The samples were dissolved in ethanol (500 µg/ml) and 0.5 mL of each solution was added to 2.5 mL Folin-Ciocalteu reagent 10% (v/v). They were then added to 2.0 mL sodium carbonate (Na₂CO₃) 4% (w/v). The absorbance was measured in a spectrophotometer at 750 nm. The standard used to construct the calibration curve was gallic acid in concentrations from 5 to 100 µg/mL. The results were expressed in mg of GAE (gallic acid equivalents) per gram of sample.

Determination of total flavonoid

The flavonoid content was determined by a colorimetric method using an aluminum chloride reagent (KALIA et al., 2008). The samples were dissolved in ethanol (1000 μ g/mL) and 0.5 mL of each solution was added to 0.1 mL aluminum chloride (AlCl₃.6H₂O) 10% (w/v) and to 0.1 mL of potassium acetate (CH₃COOK) 1M (w/v). The absorbance was measured in a spectrophotometer at 425 nm. The standard used to construct the calibration curve was quercetin in a concentration range of 5 to 100 μ g/mL. The results were expressed in mg of EQ (quercetin equivalents) per gram of sample.

Evaluation of antiradical activity

The antiradical activity was determined based on the 2,2-diphenyl-1-picrylhydrazyl (DPPH) radical scavenging method (YEN; CHANG; DUH, 2005). The samples and standards (ascorbic acid and BHT) were dissolved in ethanol (25–400 μ g/mL) and 2.0 mL of each solution was added to 0.5 mL DPPH (2, 2-diphenyl-1-picrylhydrazyl) solution (0.5 mM). The absorbance was measured at 517 nm and the results were expressed as percentages of DPPH radical scavenged.

Assessment of reducing power

The reducing power of the samples was measured according to previously described methods (YILDRIM et al., 2001a; YILDRIM et al., 2001b). The samples and standards (quercetin, ascorbic acid and BHT) were dissolved in ethanol (25–400 µg/mL). To 2.0 mL of each solution was added 2.5 mL phosphate buffer 0.2 M (pH 6.6), 2.5 mL potassium ferricyanide (K₃[Fe(CN₆)]) (1% w/v) and 2.5 mL trichloroacetic acid (10% w/v). To 2.5 mL of this mixture was added 0.5 mL ferric chloride (FeCl₃) 0.1% w/v. The absorbance was measured at 700 nm and the results were expressed as the percentages of ferric ions reduction.

Evaluation of antimicrobial activity

The dry extract and the fractions were dissolved in dimethyl sulfoxide (DMSO) (50 mg/mL) and antimicrobial activity was assessed according to the agar diffusion method according to the methodology proposed in the documents M7-A6 (CLSI, 2003) for bacteria, M24-A2 (CLSI, 2011) for mycobacteria and M44-A2 (CLSI, 2009) for fungi. The microorganisms used consisted of Gramnegative bacteria: Escherichia coli (ATCC 25922), Pseudomonas aeruginosa (ATCC 27853), Proteus mirabilis (ATCC 25922), Salmonella tiphy (ATCC 14028), Enterobacter cloacae LMI-Unifal; Grampositive bacteria: Enterococcus faecalis (ATCC 51299), Micrococcus luteus (ATCC 9341), Bacillus subtilis (ATCC 6633), Staphylococcus aureus (ATCC 6538), Bacillus cereus (ATCC 11778); Fungi: Candida albicans (ATCC 10231) and Saccharomyces cerevisiae (ATCC 2601) and mycobacteria: Mycobacterium tuberculosis (ATCC 25177) (H37Ra) and Mycobacterium bovis (BCG sample). The chlorhexidine 0.12% was used as positive control and the distilled water as negative control.

The minimum inhibitory concentration (MIC) was determined in samples that inhibited microbial growth in the agar diffusion test,

608

according to the document M27A3 (CLSI, 2008). The tests were performed in sterile 96-well microplates. The dry extract and the fractions were dissolved in DMSO to obtain initial concentration of 25 mg/mL. From this stock solution, serial dilutions were performed to achieve a concentration of 0.025 mg/mL.

Evaluation of cytotoxicity

Peritoneal murinos macrophages were maintained in RPMI 1640 to 37° C and 5% CO₂, arranged in 24 wells in the ratio $8x10^{5}$ per well, to which were added the extract and fractions to be evaluated in various concentrations (0.1 to 160 µg/mL) and incubated for 72 hours. After the incubation period were added 50 µl of MTT (3-(4,5-dimethylthiazol-2-yl)-2,5diphenyltetrazolium

bromide) to each well, with new incubation for 4 hours. The cells were lisadas with DMSO and evaluated in UV/VIS spectrophotometer Shimadzu, double-beam, 2550 model to 570 nm compared to the control without adding drugs (PEREIRA et al., 2010).

Statistical analysis

The experiments were performed in triplicate and the mean and standard deviation of the results were determined. Values were submitted to analysis of variance (ANOVA) followed by Scott and Knott (1974) test (P < 0.05). Correlation analyses were assessed using Spearman correlation coefficients (r).

RESULTS AND DISCUSSION

In chemical analyses by mass spectrometry, it was found that compounds of molecular masses 564, 578, 594 and 610 g/mol (Figure 1) are common to the extract and all fractions, but with different intensities in each. These compounds probably correspond to flavonoids that were deprotonated and recorded as their respective ions ([M-H]⁻) (GATES; LOPES, 2012). Fragmentations in multiple stages (MSⁿ) were obtained for these ions, which made it possible to propose their molecular structures based on the literature and likely fragmentation mechanisms.

Analyzing the data in MS^2 (Table 1), it was observed that the fragmentation of the ion with m/z 609 ([M-H]⁻) resulted in a base peak of m/z 447 and another peak of m/z 463, suggesting the loss of mass 162 (hexose) and 146 (6-deoxy-hexose) sugars, respectively (WOLFENDER et al. 1992; GATES; LOPES, 2012).



Figure 1. Spectrum mass in *full scan* mode, obtained by direct injection of dry extract.

The ion of m/z 593 ([M-H]⁻) showed a base peak of m/z 447, which is likely to be due to the loss of a sugar of mass 146 (6-deoxy-hexose) (WOLFENDER et al., 1992; RANA et al., 2015). The peak with m/z 301 is the aglycone quercetin (SUN et al., 2007; DEVARAJI; KRISHNA; VISWANATHA, 2011; GATES; LOPES, 2012; RANA et al., 2015; VALLVERDÚ-QUERALT et al., 2015), formed after the loss of sugars. In the MS^2 spectrum the ion of m/z 577 ([M-H]⁻) observed with a base peak at m/z 431 and another peak at m/z285 may result from the loss of one and two sugar molecules of mass 146 (6-deoxy-hexose), respectively (WOLFENDER et al.. 1992: ZANUTTO et al., 2013; RANA et al., 2015). The ion with m/z 563 ([M-H]⁻) showed a base peak of m/z 417 and another peak of m/z 430, which may be the result of loss of a sugar of mass 146 (6-deoxyhexose) and a mass 132 sugar (pentose), respectively (WOLFENDER et al., 1992). The peak with m/z 285 corresponds to the aglycone kaempferol (SUN et al., 2007; DEVARAJ; KRISHNA; VISWANATHA, 2011; RANA et al., 2015; VALLVERDÚ-QUERALT et al., 2015) formed after the loss of sugar molecules.

Generally, radishes are characterized by a high content of phenolic compounds, including the flavonoids quercetin and kaempferol (SHIN et al., 2015; KIM; BASKAR; PARK et al., 2016; RICARDO et al., 2018; IYDA et al., 2019). Other compounds such as, alkaloids, glycosinolates, pigments and proteoglycans may also be present and exhibit antimicrobial, antioxidant, antitumor and antiviral activities (ARUNA; YERRAGUNI; RAJU, 2012; SHIN et al., 2015).

Flavonoids	\mathbf{R}_1	R_2	R ₃	R_4	R_5	[M-H] ⁻	Fragments main
1	O-hex	OH	O-6-dhex	OH	OH	609	463, 447, 301
2	OH	OH	O-6-dhex	Н	O-6-dhex	593	447, 301
3	Н	OH	O-6-dhex	OH	O-6-dhex	577	431, 285
4	OH	OH	O-6-dhex	Н	pent	563	417, 430, 285

Table 1. Proposed structures of some flavonoids found in dry extract and the fractions.

Legend: hex (hexose); pent (pentose); 6-dhex (deoxy-hexose).

The phenol contents (Table 2) were higher in the fractions obtained after the concentration of the dry extract, and were most concentrated in the butanol fraction. The flavonoid contents were also higher in the fractions than in the dry extract except for the aqueous fraction, and were most concentrated in the ethyl acetate fraction. Chorol (2019) in his study with methanolic and acetonic extracts from *R.sativus* L. leaves obtained phenolic compound concentrations of 20 mg GAE/g e 3.6 mg GAE/g, respectively. Lower values than the values obtained in this work. Phenolic compounds and vitamin C are found in higher amounts in antioxidant Brassicaceae family plants (PODSEDEK, 2007). Flavonoids are reducing agents which can act to capture and neutralize oxidizing species, acting synergistically with other antioxidants such as vitamins C and E. The capacity of flavonoids to chelate metals also prevents these acting as catalysts in the formation of free radicals (SILVA; VALE; FELÍCIO, 2015). This suggests the protective role of these metabolites in reducing the risk of cardiovascular disease and other diseases associated with the formation of free radicals (PEREIRA; CARDOSO, 2012).

609

Samples	Total phenols content (mg GAE/g of sample) *	Total flavonoid content (mg QE/g of sample) *
EF	$47,02 \pm 0,6^{a}$	$20,36 \pm 0,6^{b}$
FrHex	$96,09 \pm 3,1^{\circ}$	$70,93 \pm 0,4^{\circ}$
FrAcOEt	$99,95 \pm 1,4^{\circ}$	$98,02 \pm 0,2^{e}$
FrBuOH	$121,27 \pm 1,1^{d}$	$89,87 \pm 0,6^{d}$
FrAq	$52,48 \pm 0,3^{b}$	$3,28 \pm 0,4^{a}$

Table 2. Total phenolic and flavonoid content of dry extract and fractions of leaves of *R. sativus* var. *oleiferus*

Legend: EF: dry extract of the leaves; FrHex: hexane fraction; FrAcOEt: ethyl acetate fraction; FrBuOH: butanol fraction; FrAq: aqueous fraction; GAE: gallic acid equivalents; QE: quercetin equivalents.

Evaluation of the antioxidant activity

Among the samples analyzed in this work, it was observed that the ethyl acetate fraction presented higher antioxidant capacity (Table 3), possibly due to the higher flavonoid content (Table 2). In addition, the ethyl acetate fraction presented a percentage of free radical sequestration close to the standards of quercetin and ascorbic acid, demonstrating the potential of this fraction. Goyeneche et al. (2015) demonstrated in their study that the leaves of *R. sativus* L. presented a four times higher concentration of flavonoids in relation to the root, showing an excellent antioxidant capacity. The increase of the activity as a function of the concentration of these compounds was also evidenced in the study of Park et al. (2016) carried out with three cultivars of radish. Eveline and Pasau (2019) determined that the ethyl acetate would be the best type of solvent for the extraction of bioactive compounds in radishes, due to higher flavonoid concentrations and to a higher capability of neutralization of free radicals obtained.

Table 3. Antirradicalar activity and reducing power (to $100 \,\mu g/mL$) of dry extract and fractions of leaves of *R*. *sativus* var. *oleiferus*.

Samples	Percentages of DPPH radical scavenging*	Percentages of ferric ions reduction*
EF	$26,48 \pm 0,4^{a}$	$6,49 \pm 0,28^{a}$
FrHex	$47,70 \pm 0,1^{\circ}$	$10,54 \pm 0,29^{d}$
FrAcOEt	$83,45 \pm 0,5^{g}$	$11,34 \pm 0,29^{e}$
FrBuOH	$57,20 \pm 0,3^{d}$	$10,71 \pm 0,29^{d}$
FrAq	$40,16 \pm 0,3^{b}$	$8,33 \pm 0,36^{b}$
Quercetina	$81,44 \pm 0,2^{\rm f}$	-
Ácido Áscórbico	$90,43 \pm 0,3^{h}$	$20,53 \pm 0,51^{\rm f}$
BHT	$63.7 \pm 0.10^{\rm e}$	$9,13 \pm 0,28^{\circ}$

Legend: EF: dry extract of the leaves; FrHex: hexane fraction; FrAcOEt: ethyl acetate fraction; FrBuOH: butanol fraction; FrAq: aqueous fraction; BHT: butylated hydroxytoluene.

*Means with different letters in the same column are significantly different (Scott and Knott, P < 0.05).

In previous studies with plants belonging to the Brassicaceae, the percentages of DPPH radical scavenged were found ranging from 14% to 23% in extracts of rocket (*E. sativa*) at a concentration of 0.1 mg/mL (ARBOS et al., 2010), 51.05% in broccoli extracts (*B. oleracea* var. *italica*), 29.06% in a radish extract (*R. sativus* L. var. *radicle*) (MELO; FARIA, 2014) and 57.7% in essential oils of leaves of *Eruca vesicaria* (HICHRI et al., 2019), results lower than those obtained in the ethyl acetate fraction of this study.

The reducing power method complements the antiradical activity assessment, since it evaluates

the capability of the samples in reducing ferric ions into ferrous ions by phenolic hydroxyls (ROGINSKY; LISSI, 2005). The test results confirm the samples can inhibit the oxidative damages, since the ferric ions are catalyzers of the formation of radical species (LLESUY, 2002).

Evaluation of the antimicrobial activity

The dry extract of the leaves of *R. sativus* var. *oleiferus* presented larger inhibition zones of microbial growth against *B. subtilis*, and this difference was significantly superior to those produced by the fractions (P < 0.05) (Table 4).

Table 4. The averages values of inhibition zones of microbial growth (mm) of dry extract and fra-	ctions of
leaves of <i>R. sativus</i> var. <i>oleiferus</i> evaluated by the agar diffusion method.	

	×	А	verages inhi	bition zones	(mm)*		
Microorganisms	EF	FrHex	FrAcOEt	FrBuOH	FrAq	Cl	Ag
Gram-positive							
B. subtilis	18,67 ^d	9,00 ^b	9,34 ^b	6,00 ^b	5,30 ^b	28,00 ^e	0,00 ^a
B. cereus	18,00 °	0,00 ^a	9,34 ^b	5,60 ^b	0,00 ^a	27,34 ^e	0,00 ^a
M. luteus	10,67 ^b	0,00 ^a	9,67 ^b	0,00 ^a	0,00 ^a	34,67 ^f	0,00 ^a
E. faecalis	9,34 ^b	0,00 ^a	7,34 ^b	0,00 ^a	0,00 ^a	19,67 ^d	0,00 ^a
S. aureus	13,00 °	8,00 ^b	9,34 ^b	0,00 ^a	0,00 ^a	27,34 ^e	0,00 ^a
Gram-negative							
E. coli	0,00 ^a	0,00 ^a	0,00 ^a	0,00 ^a	0,00 ^a	21,34 ^d	0,00 ^a
P. aeruginosa	0,00 ^a	0,00 ^a	0,00 ^a	0,00 ^a	0,00 ^a	21,34 ^d	0,00 ^a
P. mirabilis	0,00 ^a	0,00 ^a	0,00 ^a	0,00 ^a	0,00 ^a	17,00 ^c	0,00 ^a
S. typhimurium	0,00 a	0,00 a	0,00 a	0,00 a	0,00 ^a	21,34 ^d	0,00 ^a
E. cloacae	0,00 ^a	0,00 ^a	0,00 ^a	0,00 ^a	0,00 ^a	11,67 ^b	0,00 ^a
Fungi							
S. cerevisae	0,00 ^a	0,00 ^a	0,00 ^a	0,00 ^a	0,00 ^a	15,34 °	0,00 ^a
C. albicans	0,00 a	0,00 a	0,00 a	0,00 a	0,00 ^a	15,67 °	0,00 a
Mycobacteria							
M. tuberculosis	0,00 ^a	0,00 a	0,00 a	0,00 a	0,00 a	20,00 ^b	0,00 a
M. bovis	0,00 ^a	0,00 ^a	0,00 ^a	0,00 ^a	0,00 a	20,00 ^b	0,00 ^a

Legend: EF: dry extract of the leaves; FrHex: hexane fraction; FrAcOEt: ethyl acetate fraction; FrBuOH: butanol fraction; FrAq: aqueous fraction; Cl: Chlorhexidine 0.12%; Ag: Distilled water.

* Means with different letters in the same column are significantly different (Scott and Knott, P < 0.05).

The antibacterial activity is related to bioactive compounds present in the dry extract and the fractions of the leaves. However, which chemical compounds, alone or in combination, are active can not be determined in this study. There is a possibility that the compounds act synergistically against *B. subtilis*, *B. cereus*, *M. luteus*, *S. aureus* and *E. faecalis*, because the dry extract of the leaves produced larger inhibition zones compared to their fractions.

The antimicrobial activity of flavonoids can be attributed to their role causing changes in the physical-chemical properties of cell membranes, such as decreasing their fluidity (TSUCHIYA, 2010), activity of flavonoids can be attributed to the inhibition of nucleic acid synthesis, energy metabolism and cell wall synthesis (TIM; LAMBB, 2011).

The dry extract and the fractions of the leaves of fodder turnip proved to be active only against gram-positive bacteria. This absence of antimicrobial activity is likely due to different actions of the active compounds of the extract on the walls or cell membranes of the microorganisms, since these exhibit variations in their chemical constitutions cells. The cell wall of gram-negative microorganisms is more complex, and has an additional outer membrane with a second lipid bilayer which strongly adheres to the peptidoglycan layer, giving greater stiffness and antigenicity (GUIMARÃES; MOMESSO; PUPO, 2010). The high concentration of high molecular weight lipids present in the cell wall of mycobacteria acts as a barrier for polar compounds (ARANTES et al., 2005). The cytoplasmic membranes of fungi include steroids, lipids and also proteins (TORTORA; FUNKE; CASE, 2012). This may explain the microorganisms resistance of these to hydroethanolic extracts. However, several studies have demonstrated the antimicrobial capacity of different extracts of the plant against some of these microorganisms (SHUKLA et al., 2011; AHMAD et al.. 2012: UMAMAHESWARI; AJITH: ASOKKUMAR, 2012; JANJUA; SHAHID: ABBAS, 2013; CHIHOUB et al., 2019; DUY et al., 2019; IYDA et al., 2019), suggesting that future experiments with extracts of the fodder turnip of different polarities are promising.

The lowest MIC values (Table 5) were obtained against *M. luteus*, and the ethyl acetate fraction showed the lowest MIC against this microorganism, followed by the aqueous and butanol fractions. The ethyl acetate fraction was shown to be more effective, with MICs between 0.1 and 12.5 mg/mL.

	U U	Minimum Inł	nibitory Concent	ration (mg/mL)	
Bacteria	EF	FrHex	FrAcOEt	FrBuOH	FrAq
B. subtilis	3,12	6,25	0,4	3,12	3,12
B. cereus	12,5	3,12	0,8	1,56	3,12
M. luteus	12,5	nd	0,1	0,4	0,2
E. faecalis	12,5	12,5	12,5	12,5	12,5
S. aureus	6,25	nd	0,1	3,12	3,12

Table 5. Determinations of Minimum Inhibitory Concentration (MIC) (mg/mL) of dry extract and fractions of leaves of R. sativus var. oleiferus.

Legend: EF: dry extract of the leaves; FrHex: hexane fraction; FrAcOEt: ethyl acetate fraction; FrBuOH: butanol fraction; FrAq: aqueous fraction; nd: not detected at maximum concentration tested (12.5 mg/mL).

Some techniques used for antimicrobial evaluation may also lead to false negative results. The agar diffusion test, for instance, is only efficient for polar substances, allowing the diffusion of these through the culture medium. In addition, factors such as the presence of bacterial enzymes, the composition of the medium, the diffusion of the substance in the medium, the inoculum density, the time and temperature of incubation, the stability of substances and the molecular mass may hinder diffusion in the medium (SILVEIRA et al., 2009), which can prevent the contact of the extracts with microorganisms, leading to negative results. However, the plate microdilution assay allows direct contact of the extract with the microorganisms, avoiding possible limitations of the agar diffusion method.

The extracts that present MIC less than 100 µg/mL can be considered to have good antimicrobial activity, MIC between 100-500 µg/mL are moderately active, MIC between 500-1000 µg/mL are very active and higher MIC than 1000 µg/mL are considered inactive (HOLETS et al., 2002; FABRY et al., 2008). Thus, from table 5,

Reducing power method x Flavonoids

it can be seen that ethyl acetate fractions, followed by aqueous and butanol fractions showed the best antimicrobial activity, as demonstrated by their MIC in the range $100-800 \mu g/mL$.

In a study with aqueous extracts of the fruits of R. sativus L. MIC values of 5 mg/mL against S. aureus were found (EDZIRI et al., 2012), higher than those obtained in the ethyl acetate, butanol and aqueous fractions of the leaves of R. sativus var. oleiferus. Jadoun et al. (2016) obtained MIC values of 0.5 mg/mL for the same microorganism, however using sulfur compounds isolated from seeds of R. sativus L. Chihoub et al. (2019) obtained MIC of 20 mg / mL for E. faecalis using hydroethanolic extracts 80% of R. sativus L leaves. Therefore, the results obtained highlight the fodder turnip as a source of antibacterial compounds.

Correlation analyze

The results of the correlation analyze between the main constituents detected and the biological activities (Table 6) showed values of positive r, according to the classification of Callegari-Jaques (2003).

Table 6. Statistical analysis of the correlation between the parameters evaluated.					
Correlation	r **				
Antimicrobian activity					
MIC x Phenolic	0,40				
MIC x Flavonoids	0,33				
Antioxidant activity "in vitro"					
DPPH• method x Phenolic	0,80				
DPPH• method x Flavonoids	0,77				
Reducing power method x Phenolic	0,70				

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Legend: r = correlation coefficient; MIC = Minimum Inhibitory Concentration; DPPH• = radical DPPH.

**According to the classification of Callegari-Jacques (2003): 0.00 <r < 0.30 shows weak correlation; 0.30 <r < 0.60, shows moderate correlation; 0.60 <r <0.90, shows strong correlation and 0.90 <r <1.00 shows a strong correlation between the evaluated parameters.

0,64

The analyses showed positive correlations the lowest minimum between inhibitory concentration values obtained and the highest amounts of phenols and flavonoids in the extracts and fractions (Table 6). This can be considered as a moderate to correlation (CALLEGARI-JACOUES, 2003) between the presence of these metabolites and the best antimicrobial activity values. This indicates that phenolic compounds, particularly flavonoids, have an important role in the antimicrobial activity of the dry extract and the fractions of leaves of R. sativus var. oleiferus. However it is likely that the activity is not only related to this class of secondary metabolites.

The analysis of the correlation between the highest values in the antiradical activity measured by DPPH radical scavenging and the highest reducing power with the highest amounts of total phenols and flavonoids in the dried extract and the evaluated fractions also showed positive r values there is a strong (Table correlation 6), (CALLEGARI-JACQUES, 2003) between these metabolites and the highest values of antioxidant activity in vitro. This shows that phenolic compounds, especially flavonoids, are the main compounds responsible for the antioxidant activity. Similar correlations were obtained in the study by Li et al. (2018) with 12 cruciferous vegetables.

Chihoub et al. (2019) in their study with 80% of hydroethanolic extract from *R. sativus* L. leaves concluded that the phenolic compounds are strongly correlated to the antioxidant and the antimicrobial activity.

Cytotoxicity

The extract and fractions not shown cytotoxic for peritoneal murinos macrophages until the highest concentration evaluated (160 μ g/mL). A positive aspect in studies of cytotoxic activity appears in search of potentially anticancer substances. Pocasap, Weerapreeyakul and Barusrux (2013) evaluated the antitumor effect of the extract of *R. sativus* var. *caudatus* against a colon cancer cell line HCT116, by MTT assay and the results showed high citotoxicity against extract these cells, with CC50 of 9.42 ± 0.46 μ g/ml. To view more cytotoxic trials with the species *R. sativus* var. *oleiferus* can lead to relevant results.

CONCLUSION

The potential antioxidant and antimicrobial actions of the extract and the fractions of leaves of *R. sativus* var. *oleiferus*, and the ethyl acetate fraction was promising for further studies. Moreover, it revealed that concentrations of the major components detected showed positive correlations with the biological activities evaluated. The samples did not show cytotoxicity until the highest concentration evaluated.

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RESUMO: O rabanete (*Raphanus sativus* L.) é um vegetal da família Brassicaceae cultivado em todo o mundo e possui diversas propriedades medicinais. Suas atividades biológicas estão relacionadas aos vários metabólitos secundários presentes na espécie, especialmente os compostos fenólicos. Desta forma, os objetivos deste estudo foram realizar análises químicas e avaliar as atividades antioxidante e antimicrobiana do extrato seco e das frações das folhas de R. sativus var. oleiferus Metzg. As amostras foram analisadas em espectrômetro de massas e o potencial antioxidante foi avaliado pelos métodos do radical DPPH (2,2-difenil-1picrilhidrazila) e do poder redutor. A atividade antimicrobiana foi determinada pelos métodos de difusão em ágar e da microdiluição. Observou-se que os fenóis totais se concentraram na fração butanólica (121,27 mg EAG/g), enquanto que e os teores de flavonoides concentraram-se na fração acetato de etila (98,02 mg EQ/g). A fração acetato de etila apresentou os melhores resultados antioxidantes, com porcentagem de sequestro dos radicais DPPH de 83,45% e com porcentagem de redução dos íons férrico de 11,34%. A análise da atividade antimicrobiana revelou que o extrato seco teve maior média de halos de inibição frente ao Bacillus subtilis (18,67 mm). Os menores valores da concentração inibitória mínima foram para Micrococcus luteus, sendo que a fração acetato de etila demonstrou menor concentração inibitória mínima (0,1 mg/mL) para esse microorganismo. Houve uma forte correlação entre a atividade antioxidante e o teor de fenóis e de flavonoides. Os resultados demonstraram potenciais ações antioxidante e antimicrobiana do extrato e das frações avaliados, sendo a fração acetato de etila promissora para estudos posteriores.

PALAVRAS-CHAVE: Nabo forrageiro. Brassicaceae. Compostos fenólicos. Plantas Medicinais.

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