

## Ag/SiO<sub>2</sub> nanocomposite mediated by *Escherichia coli* D8 and their antimicrobial potential

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### Article info

#### Article history:

Received: 20<sup>th</sup> June 2021

Accepted: 8<sup>th</sup> November 2021

#### Keywords:

Antimicrobial

Biosynthesis

*Escherichia coli*

Nanocomposite

Silica

Silver

### Abstract

Silica (SiO<sub>2</sub>) has a fundamental role in the recuperation of plants in response to environmental stresses, besides the induction of resistance against plant diseases. Silver nanoparticles (AgNPs) have a superior antimicrobial activity. The combination between SiO<sub>2</sub> and AgNPs is a promising approach due to their antimicrobial activity, biological activity, low toxicity, and high stability of the produced nanocomposite. The current study postulated a green method for silver/silica nanocomposite (Ag/SiO<sub>2</sub>NC) synthesis at room temperature using the crude metabolites of *Escherichia coli* D8 (MF062579) strain in the presence of sunlight. UV-Vis spectrophotometry, X-ray diffraction (XRD), Fourier transform-infrared spectroscopy (FTIR), and transmission electron microscopy (TEM) analyses have characterized the biosynthesized nanocomposite. TEM study of Ag/SiO<sub>2</sub>NC showed an average particle size of ~32 – 48 nm whereas AgNPs showed a mean size of 18 – 24 nm. The negative charged Ag/SiO<sub>2</sub>NC (-31.0 mV) showed potent antimicrobial activity against *Bacillus cereus* ATCC6633, *Klebsiella pneumoniae* ATCC33495, *Staphylococcus aureus* (ATCC25923), *E. coli* (ATCC25922), *Candida albicans* (ATCC10231), and *Botrytis cinerea* (Pers: Fr.). The minimum inhibitory concentration (MIC) test showed a dose-dependent manner of Ag/SiO<sub>2</sub>NC antimicrobial action. MIC values of Ag/SiO<sub>2</sub>NC against the tested pathogens exhibited 125 and 6.25 µg.mL<sup>-1</sup> as antibacterial and antifungal agents, respectively. TEM micrographs showed changes in the pathogens treated with Ag/SiO<sub>2</sub>NC including wrinkling, damage, and rupture of the bacterial cell membrane. In addition, the formation of a mucilage matrix connecting the hyphal cells, the appearance of big vacuoles and lipid droplets with severe leakage of cytoplasmic contents of the treated *B. cinerea* were also recorded.

### Introduction

Nowadays, there is a tendency to use materials after converting them into their nano-form, because of the new and promising advantages and unique properties gained in the new nano-form such as antimicrobial activity, chemical, magnetic, electronic, or mechanical properties because of the change of quantum and surface boundary effects compared with their bulk materials (Chhipa and

Joshi 2016; Musere *et al.* 2021). Nanomaterials can be described as materials with a size of 1 – 100 nm, known as the nano-scale range. Synthesis of nanometals can be done using different methods such as chemical, physical or biological techniques. Conventional hypothesis of chemical synthesis of nanoparticles (NPs) might possess several serious problems due to using expensive toxic chemicals (Zhang *et al.* 2021).

Physical methods include high radiation and stabilizing agents that might be dangerous for human health and to the environment (Awwad *et al.* 2020). Accordingly, the green synthesis of NPs by plant extracts or microbial metabolites plays an important role in the reduction of metal ions into NPs and capping them in supporting their stability (Parveen *et al.* 2016). Biological NPs have different characters compared with chemical or physical NPs, with superior stability and suitable dimensions due to the one-step technique (Narayanan and Sakthivel 2011).

Silver nanoparticles (AgNPs) were used as a potent antimicrobial agent during the last decades (Hamad *et al.* 2020). It showed antimicrobial activity against pathogens such as *Escherichia coli* (Yang *et al.* 2021), *Staphylococcus aureus* (Enan *et al.* 2021), *Klebsiella pneumoniae* (Pareek *et al.* 2021), *Candida albicans* (Takamiya *et al.* 2021), and *Botrytis cinerea* (Ouda 2014). The aggregation of AgNPs is a common problem that decreases their biological activity (El-Dein *et al.* 2021). One way to enhance the metal NPs stability is by stabilizing them by embedding them inside a polymer, which prevents their aggregation, even at high-volume fractions (Baheiraei *et al.* 2012). Several previous studies have focused on the synthesis of stable monodisperse silica-coated with nanometals, mainly Ag and gold (Au) (Chen *et al.* 2017; Si *et al.* 2019; Li *et al.* 2021). Silica (SiO<sub>2</sub>) can act as the platform for developing NPs moreover having antimicrobial properties owing to their large surface area, positive surface charge, and monodispersity. Gankhuyag *et al.* (2021) reported that SiO<sub>2</sub> might increase the stability of the nanometals and prevent their aggregation. In addition, the net positive charge of SiO<sub>2</sub> facilitates a greater number of AgNPs to interact with the negatively charged surface of bacteria, resulting in highly efficient antimicrobial activity (Jayasuriya 2017). Egger *et al.* (2009) and Sohrabnezhad *et al.* (2020) demonstrated the antibacterial activity of silver/silica nanocomposite (Ag/SiO<sub>2</sub>NC) against both *E. coli* and *S. aureus*. Ag/SiO<sub>2</sub>NC revealed marked changes in the bacterial cell contents, including the cell wall integrity, metabolism, and genetic stability of *Pseudomonas aeruginosa* (Anas *et al.* 2013). Also, Xu *et al.* (2009) reported the antibacterial effects of Ag/SiO<sub>2</sub> core-shell particles against *E. coli* and *S. aureus*. Ag/SiO<sub>2</sub>NC had

antifungal potential against *B. cinerea* as reported by Oh *et al.* (2006). Youssef and Roberto (2021) demonstrated the antifungal activity of chitosan/silica nanocomposite against *B. cinerea*. Ag/SiO<sub>2</sub>NC showed fungicidal activity against the pathogenic fungi in the soybean plants (*Fusarium oxysporum* and *Rhizoctonia solani*) as reported by Nguyen *et al.* (2016) and *Aspergillus flavus* as reported by Diagne *et al.* (2020). Hence, new distinctive structures of Ag/SiO<sub>2</sub>NC could present a new prospect for its antimicrobial activity.

The present study aimed to evaluate the ability of the crude metabolite of *E. coli* D8 (MF062579) for reducing the silver nitrate (AgNO<sub>3</sub>) into AgNPs extracellularly and also their binding with SiO<sub>2</sub> in a new one-step green approach. The antimicrobial potential of Ag/SiO<sub>2</sub>NC was studied against some pathogenic strains, comparing their activity to the standard commercial antibiotics.

## Experimental

### Microbial cultures

*E. coli* D8 (AC: MF062579) and the pathogenic bacterial and fungal strains were obtained from the culture collection of Botany and Microbiology Department, Faculty of Science, Damietta University, Egypt.

### Chemicals

The chemicals included silver nitrate (Panreac Quimica S.L.U, Barcelona, Spain), silica (Silicon dioxide nanoparticles, particle size 190 – 250 nm, mesoporous, pore size 4 nm, Sigma-Aldrich), culture media, and other chemicals (Sigma Aldrich Chemical Pvt. Ltd., India). Penicillin G potassium (Buffered Pfizerpen) and fluconazole (Diflucan) were purchased from Pfizer Inc., New York, NY.

### Biosynthesis of silver nanoparticles and silver/silica nanocomposite

Silver nanoparticles were prepared according to El-Zahed *et al.* (2021). In brief, *E. coli* D8 agar slants were sub-cultured on nutrient agar plates (37 °C, 24 h). Then the grown colonies were inoculated into a nutrient broth medium with 0.5 McFarland standard ( $1 - 2 \times 10^8$  CFU.mL<sup>-1</sup>) and incubated at

37 °C/150 rpm for 48 h. Later, the cell-free metabolites of *E. coli* D8 were collected by centrifugation (3H24RI intelligent high-speed refrigerated centrifuge, Herexi Instrument, and Equipment Co., Ltd) at 8,000 rpm for 20 min and filtration through a sterile 0.22 µm syringe filter (Millex GV, Millipore). For the synthesis of AgNPs, 1.5 mM of AgNO<sub>3</sub> solution was mixed with cell-free metabolites (1 % v/v) at room temperature and sunlight. For the synthesis of Ag/SiO<sub>2</sub>NC, 0.5g of AgNO<sub>3</sub> was dissolved into 50 mL of distilled water and then added to another beaker that included 100 g of SiO<sub>2</sub>. At room temperature, the previous solution was mixed well with 20 mL of *E. coli* D8 cell-free metabolites in the presence of sunlight.

The first indicator for the AgNPs and nanocomposite (NC) formation was the color change from colorless (AgNO<sub>3</sub>) or white (SiO<sub>2</sub>) into brown. After 20 min, the AgNPs and Ag/SiO<sub>2</sub>NC were collected separately by centrifugation at 10,000 rpm for 15 min several times and then dried in an oven at 50 °C for 24 h. Then, the NPs and NC were dried at 185 °C for 5 h (Sadeghi *et al.* 2013).

#### Characterization of silver/silica nanocomposite

Silver/silica nanocomposite spectra were scanned by UV/VIS/NIR Spectrophotometer (V-630, JASCO Corporation, Japan). The X-ray diffraction (XRD) pattern of the Ag/SiO<sub>2</sub>NC was performed at 2θ values (λ = 1.54 Å in the range 10 – 80 °) using a Cu X-ray tube at 40 kV and 30 mA with the X-ray diffractometer (model LabX XRD-6000, Shimadzu, Japan). Fourier transform infrared spectroscopy (FTIR) spectrum of the Ag/SiO<sub>2</sub>NC was recorded by JASCO FT/IR-4100typeA in the 400 – 4,000 cm<sup>-1</sup> range. The size and morphology of AgNPs and Ag/SiO<sub>2</sub>NC were investigated by TEM (JEOL, JEM-2100, Japan) at an accelerating voltage of 200 kV and using a carbon-coated copper grid (Type G 200, 3.05 µm diameter, TAAP, USA). Charge of AgNPs and size distribution by volume were recorded by Zeta Potential Analyzer (Malvern Zetasizer Nano-ZS90, Malvern, UK).

#### Antimicrobial potential

The antimicrobial potential of Ag/SiO<sub>2</sub>NC was tested against Gram-positive bacteria (*S. aureus* ATCC25923 and *Bacillus cereus* ATCC6633), Gram-negative bacteria (*E. coli* ATCC25922 and *K. pneumoniae* ATCC33495), yeast (*C. albicans* ATCC10231), and the phytopathogenic fungus (*B. cinerea* Pers: Fr.) by agar well diffusion and broth dilution methods. The bacterial, yeast and fungal strains were grown and tested using Mueller Hinton agar (MHA), bacto-casitone agar, and potato dextrose agar (PDA) plates, respectively. 200 µL of 0.5 McFarland standard (1 – 2 × 10<sup>8</sup> CFU.mL<sup>-1</sup>) of microbial suspension was used as an initial inoculum for each test.

#### Agar well diffusion method

Agar well diffusion assay was performed *in vitro* against the microbial strains according to the guidelines of the Clinical and Laboratory Standards Institute (Clinical and Laboratory Standards 2006). About 200 µL of 150 µg.mL<sup>-1</sup> of SiO<sub>2</sub>, Ag/SiO<sub>2</sub>NC, AgNO<sub>3</sub>, penicillin G potassium (antibacterial), and fluconazole (antifungal) were prepared and added separately into small wells (5 mm diameter of size) that were made into the solidified agar plates. Plates were incubated at 37 °C for 48 h, 30 °C for 48 h, or 28 °C for 5 days, for bacteria, yeast, and fungi, respectively. After the incubation period, inhibition zones were measured in millimeters (mm).

#### Broth dilution method

Mueller Hinton, bacto-casitone and potato dextrose broth media test tubes were prepared, autoclaved, and inoculated by 100 µL of microbial suspensions (0.5 McFarland standard (1 – 2 × 10<sup>8</sup> CFU.mL<sup>-1</sup>)) in three sets of test tubes containing different dosages of Ag/SiO<sub>2</sub>NC and Penicillin G potassium (antibacterial) or fluconazole (antifungal) concentrations (6.25 – 125 µg.mL<sup>-1</sup>). Then, the tubes were incubated at 37 °C/120 rpm for 24 h, 30 °C/120 rpm for 24 h, or 28 °C/120 rpm for 5 days, for bacteria, yeast, and fungi, respectively. The minimal inhibition concentration (MIC) for the tested pathogenic strains was determined by measuring their growth spectrophotometrically at

600 nm against negative controls (exclusive of Ag/SiO<sub>2</sub>NC). The growth inhibition percentage was calculated using the following formula (Eq. 1):

$$\% \text{ Growth inhibition} = \left[ \frac{\text{ODc} - \text{ODt}}{\text{ODc}} \times 100 \right] \quad (1)$$

where the negative control (broth media exclusive of Ag/SiO<sub>2</sub>NC) optical density; OD<sub>c</sub> and the Ag/SiO<sub>2</sub>NC-treated tested sample optical density; OD<sub>t</sub> (Clinical and Laboratory Standards 2008; 2017).

### Ultrastructural study

The ultrastructure of Ag/SiO<sub>2</sub>NC treated *E. coli* and *B. cinerea* was studied with TEM (JEOL, JEM-2100, Japan, 200kV) according to Bozzola (2007). The tested strains were subjected to Ag/SiO<sub>2</sub>NC (MIC, 6.25 μg.mL<sup>-1</sup>) for 2 h and compared with untreated bacteria and fungi as controls. The samples were fixed in 2.5 % glutaraldehyde in 0.1M cacodylate buffer at pH 7.0 and then post-fixed in 1 % osmium tetroxide, dehydrated with a graded series of ethanol, embedded in a plastic resin, and sectioned on an ultramicrotome. Ultrathin sections were double-stained with uranyl acetate and lead citrate and then loaded on carbon-coated copper grids (Type G 200, 3.05 μM diameter, TAAP, U.S.A.).

### Statistical analysis

SPSS software version 18 was used for all the statistical analysis. All values in the experiments were expressed as the mean ± standard deviation (SD) and were analyzed with a one-way Analysis of Variance (ANOVA) with a significant level set at  $P < 0.05$ .

## Results and Discussion

### Synthesis and characterization of Ag/SiO<sub>2</sub>NC

Synthesis and characterization of Ag/SiO<sub>2</sub>NC have attracted the attention of the materials community because of their promising properties (Zaferani 2018). The green synthesis approach of those NCs with controllable size and properties has applications in miniaturized catalysts, photonics,

optical devices, medical applications moreover it could be used as a potential nanomicrobicide and nanoscale growth regulator in agriculture (Das *et al.* 2019). Endless progression of microbial antibiotic-resistant mechanisms claims continuous searching for alternative approaches to deal with their risk to humans and plants (Rai *et al.* 2012). The present study provided a green approach for the synthesis of antimicrobial Ag/SiO<sub>2</sub>NC mediated by the cell-free metabolites of *E. coli* D8. FTIR spectrum confirmed the presence of proteins during the bio-reduction process. These proteins, in the *E. coli* D8 metabolite, might be including the reducing enzymes and/or some redox agents such as sulfur-containing proteins resulting in the bio-reduction of silver ions (Ag<sup>+</sup>) into AgNPs (Krishnaraj *et al.* 2012). Also, quinones (menaquinone, demethylmenaquinone, and ubiquinone) found in the *E. coli* D8 metabolite act as an electron shuttle compound and reduced Ag<sup>+</sup> into AgNPs in the presence of sunlight as reported by Duan *et al.* (2015) and Sharma *et al.* (2012).

The biosynthesis of AgNPs was confirmed through visual observation of the color change of the mixture into brown color producing an obvious absorption peak at 430 nm (Fig. 1). The brown color is because of the excitation of the AgNPs surface plasmon resonance (El-Dein *et al.* 2021). Granbohm *et al.* (2018) found the UV-Vis spectra of Ag/SiO<sub>2</sub>NC powders showed the silver SPR peak at 410 nm.

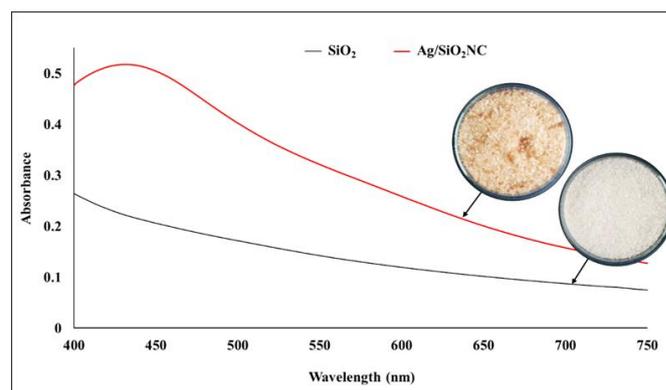
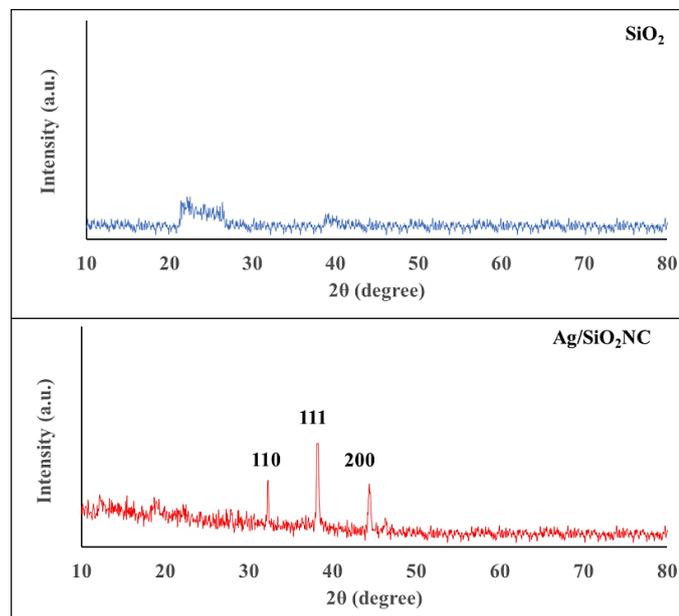


Fig. 1. The UV-Vis spectra of SiO<sub>2</sub> and Ag/SiO<sub>2</sub>NC.

The XRD patterns of SiO<sub>2</sub> were examined and showed an amorphous SiO<sub>2</sub> characteristic diffraction peak at 22.4°. Ag/SiO<sub>2</sub>NC XRD pattern revealed peaks at 2θ angles of 32.25°, 38.25° and 44.4° corresponding to the reflections of (110), (111) and (200) crystalline planes of the face-

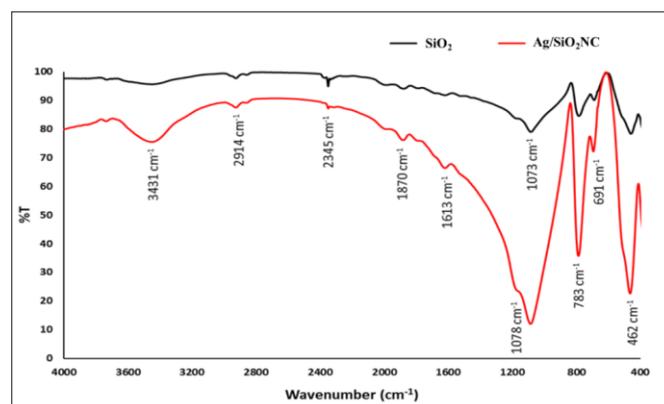
centered cubic (FCC) structure of AgNPs (Fig. 2). Also, we had found no other diffraction peaks for silver oxide in the Ag/SiO<sub>2</sub>NC XRD pattern which showed the coverage of the NC with pure AgNPs (Xu *et al.* 2015).



**Fig. 2.** The XRD patterns of SiO<sub>2</sub> and Ag/SiO<sub>2</sub>NC.

The FTIR spectra of Ag/SiO<sub>2</sub>NC were analyzed in the region of 400 – 4,000cm<sup>-1</sup> (Fig. 3). The vibration bands around 3,431 and 1,613 cm<sup>-1</sup> are attributed to the OH and carbonyl group (C=O), respectively. These signals clearly confirm the presence of bacterial compounds bounded on the surface of Ag/SiO<sub>2</sub>NC that affect protection and stability of the NC. The intense peaks around 3,431 and 2,914 cm<sup>-1</sup> are attributed to the primary and secondary amines vibrations bands, respectively. The stretch C-N vibration of aliphatic amines existed at 1,078 cm<sup>-1</sup> bands. These signals confirmed the presence of proteins in the Ag/SiO<sub>2</sub>NC synthesis. Water bands were appeared at around 1,613 cm<sup>-1</sup> corresponding to bending vibrations indicating the hygroscopic character of the powdered samples (Singh and Ahmed 2012). Si–O–Si and Si–OH absorptions bands have been observed at 1,078; 783; and 462 cm<sup>-1</sup>. The Si–O–Ag linkages stretching were also seen at around 691 cm<sup>-1</sup>. The band appears in the Ag/SiO<sub>2</sub>NC suggesting bonding between the AgNPs and the oxygen bonded to SiO<sub>2</sub>. The peaks mV, respectively). Different studies (Verma and Stellacci 2010; Anas *et al.* 2013) have studied the

450 – 800 cm<sup>-1</sup> are probably related to the pseudo-lattice vibrations (Mathur *et al.* 2006).



**Fig. 3.** The FTIR spectra of SiO<sub>2</sub> and Ag/SiO<sub>2</sub>NC.

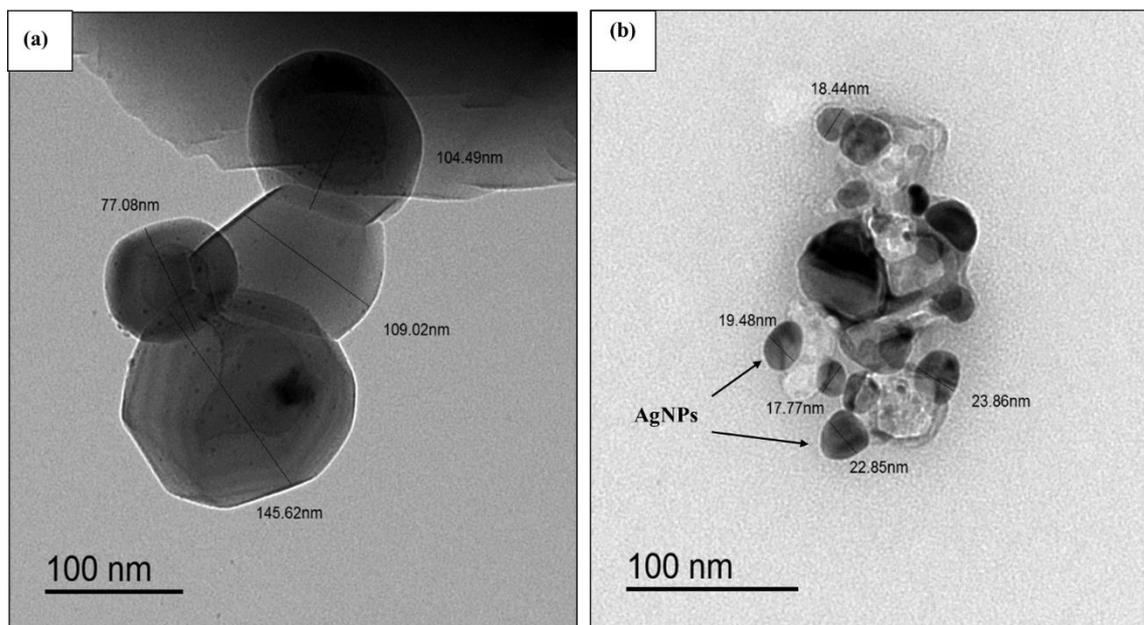
The Ag/SiO<sub>2</sub>NC was examined by the TEM (Fig. 4) to investigate the morphology and size of the AgNPs (Fig. 4B). AgNPs are embedded within the matrix and on the surface of the SiO<sub>2</sub>. TEM image showed small spherical shaped AgNPs having a diameter between 18 – 24 nm. Gu *et al.* (2011) reported that the AgNPs average particle size on the surfaces of SiO<sub>2</sub> had a little increase from 10 to 25 nm as reaction temperature increased. This should be attributed to the higher reduction rate of Ag<sup>+</sup> at the elevated reaction temperature.

SiO<sub>2</sub> has a net positive charge, while Ag/SiO<sub>2</sub>NC may have a positive or negative surface charge depending on the surface functional group and solution pH (Jana *et al.* 2007; Jayasuriya 2017). The synthesis of Ag/SiO<sub>2</sub>NC included binding primary amines as confirmed by the FTIR analysis. The primary amines were deprotonated during the bio-reduction process, were leading to a gradual decrease in the surface positive charge of SiO<sub>2</sub> and might approach zero (Jana *et al.* 2007). On the other hand, the binding between the AgNPs which are capped with highly negative proteins (El-Dein *et al.* 2021), and SiO<sub>2</sub> to give negatively charged Ag/SiO<sub>2</sub>NC. The biosynthesized Ag/SiO<sub>2</sub>NC had a negative charge, -31.0 mV (Fig. 5), which matched with Shanthil *et al.* (2012) results, -33 ± 2 mV and was better than Zhao *et al.* (2016) and El-Sheshtawy *et al.* (2020) results (-16.10 mV and -15

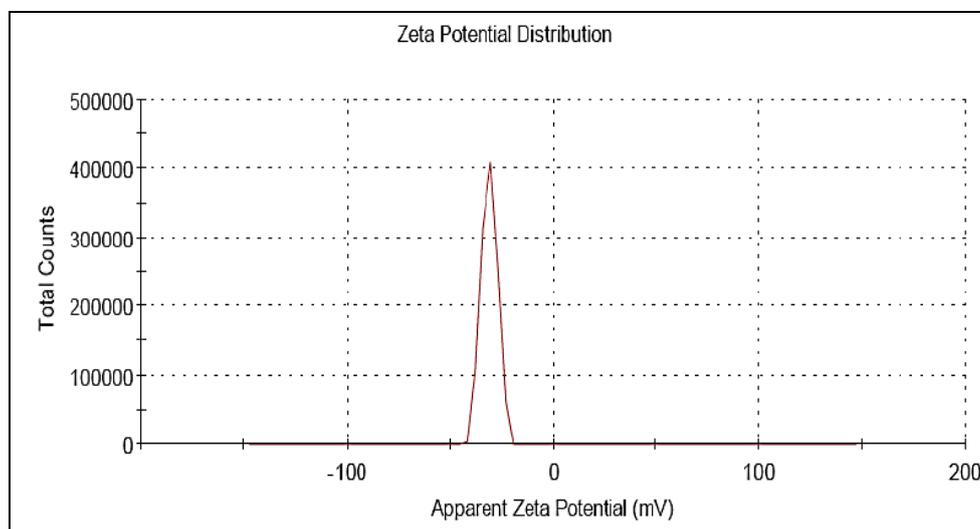
interaction of charged nanomaterials with cells showing that positively charged nanomaterials have

the greatest efficacy in penetrating the cell membrane. Other studies (Fuller *et al.* 2008; Martin *et al.* 2008) have examined the cellular uptake of negatively charged nanomaterials and proposed that negatively charged nanomaterials generate reactive oxygen species (ROS) contributes towards potent bacterial toxicity (Ivask *et al.* 2010;

Agnihotri *et al.* 2013). A further study of the synthesized Ag/SiO<sub>2</sub>NC should be taken into account to improve the antimicrobial efficacy of Ag/SiO<sub>2</sub>NC to have a positive surface charge. The positive charge of the nanomaterials increases the efficient electrostatic interaction with the negative charges of the microbial cell wall (Li *et al.* 2011).



**Fig. 4.** (a) TEM micrograph of SiO<sub>2</sub>. (b) TEM micrograph of Ag/SiO<sub>2</sub>NC.



**Fig. 5.** Zeta potential measurement analysis of Ag/SiO<sub>2</sub>NC (-31.0 mV).

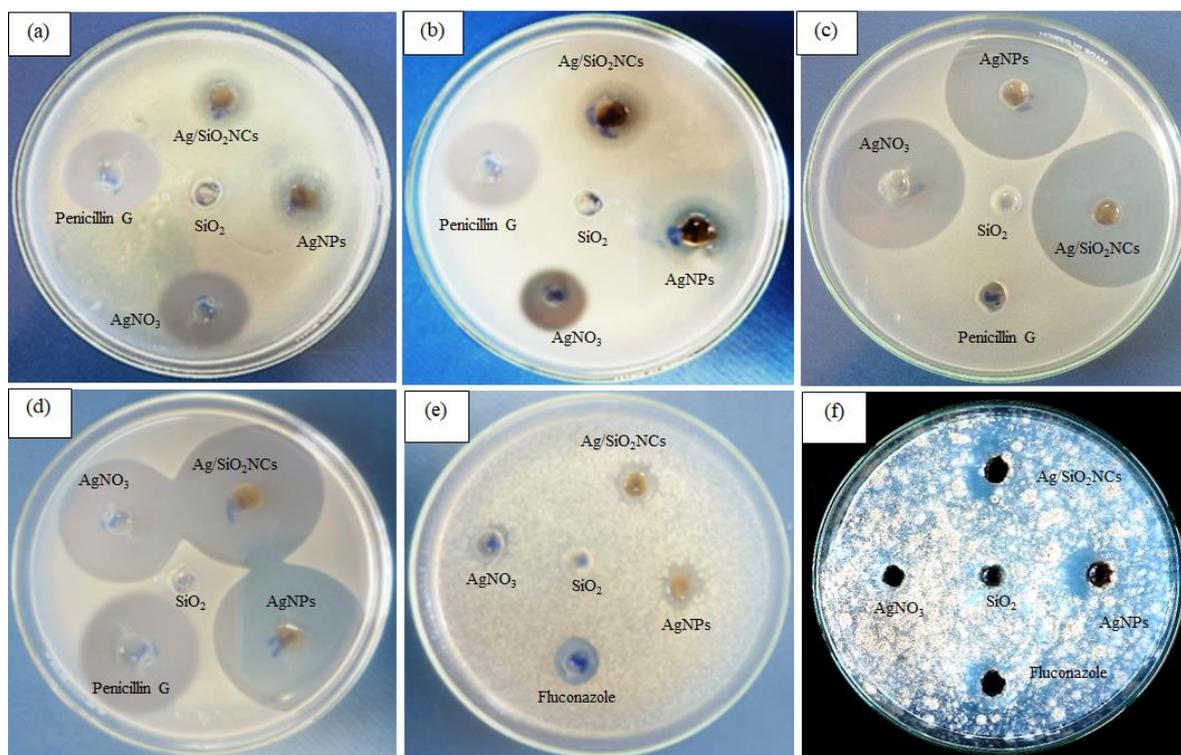
#### Antimicrobial potential of Ag/SiO<sub>2</sub>NC

The pathogenic bacteria, yeast, and fungi appeared to be more tolerant to SiO<sub>2</sub> than Ag/SiO<sub>2</sub>NC. In this study, Ag/SiO<sub>2</sub>NC was investigated to determine its antimicrobial action (Fig. 6 and Table 1). The

NC revealed very good antimicrobial potential against a wide range of microorganisms such as *K. pneumoniae*, *S. aureus*, and *B. cinerea*. The inhibition of microbial growth due to surface contact with the SiO<sub>2</sub> nanocomposite containing AgNPs demonstrated that NC functionalized with

the AgNPs has better antimicrobial action than bulk SiO<sub>2</sub>. The antibacterial potential results of Ag/SiO<sub>2</sub>NC in He et al. (2012) study revealed that Ag/SiO<sub>2</sub>NC were sensitive to *S. aureus* and *E. coli* and with the inhibition zone diameter 15.3 mm and 10.4 mm, respectively. Lu et al. (2017) studied the combination between chlorhexidine and Ag/SiO<sub>2</sub>NC and recorded that combination might produce synergistic bactericidal and candidacidal

effects and improve the microbicidal efficiency. In addition, Ag/SiO<sub>2</sub>NC showed antifungal potential against *B. cinerea* besides the antibacterial and anticandidal actions. Rodríguez-Cutiño et al. (2018) confirmed the antimicrobial properties of Ag/SiO<sub>2</sub>NC against bacteria such as *E. coli*, *B. cereus*, *S. typhimurium*, and *S. aureus* in addition to the green squash fungi: *B. cinerea* and *R. solani*.



**Fig. 6.** Antimicrobial activity of SiO<sub>2</sub>, AgNPs, and Ag/SiO<sub>2</sub>NC; (a) *B. cereus*, (b) *E. coli*, (c) *K. pneumoniae*, (d) *S. aureus*, (e) *C. albicans*, and (f) *B. cinerea*.

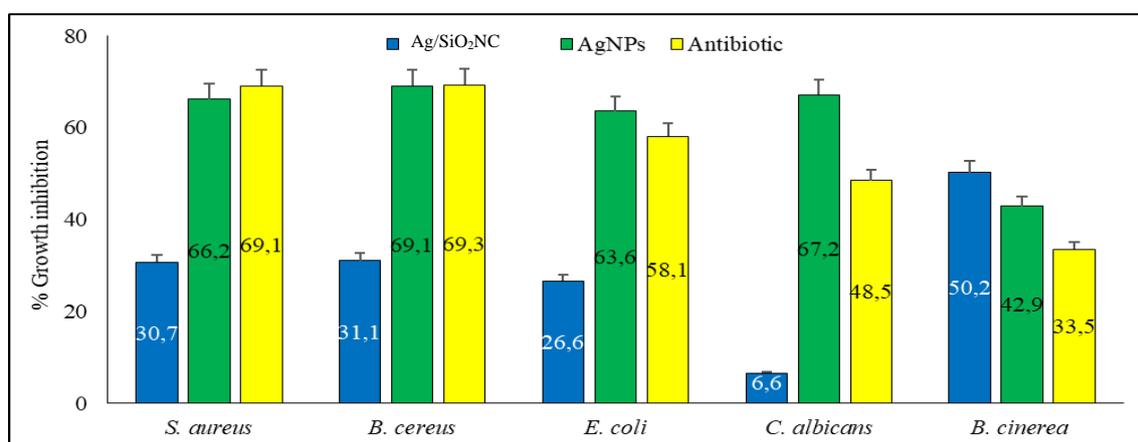
**Table 1.** Antimicrobial activity of SiO<sub>2</sub>, AgNPs and Ag/SiO<sub>2</sub>NC against the pathogenic microbial strains (Highly significant = \**P* < 0.05; *n* = 3).

| <b>Antibacterial activity (Inhibition zone, mm ± SD)</b> |                           |                          |                             |                         |
|----------------------------------------------------------|---------------------------|--------------------------|-----------------------------|-------------------------|
| <b>Substance</b>                                         | <b><i>B. cereus</i></b>   | <b><i>E. coli</i></b>    | <b><i>K. pneumoniae</i></b> | <b><i>S. aureus</i></b> |
| AgNO <sub>3</sub>                                        | 24 ± 0.06*                | 26 ± 0*                  | 34 ± 0*                     | 29 ± 0*                 |
| SiO <sub>2</sub>                                         | -ve                       | -ve                      | -ve                         | -ve                     |
| AgNPs                                                    | 30 ± 0.14*                | 30 ± 0.14*               | 38 ± 0*                     | 37 ± 0*                 |
| Ag/SiO <sub>2</sub> NC                                   | 20 ± 0.06*                | 22 ± 0.14*               | 36 ± 0*                     | 34 ± 0*                 |
| Penicillin G potassium                                   | 29 ± 0*                   | 30 ± 0*                  | -ve                         | 26 ± 0.06*              |
| <b>Antifungal activity (Inhibition zone, mm ± SD)</b>    |                           |                          |                             |                         |
| <b>Substance</b>                                         | <b><i>C. albicans</i></b> | <b><i>B. cinerea</i></b> |                             |                         |
| AgNO <sub>3</sub>                                        | 13 ± 0.06*                | 13 ± 0.14*               |                             |                         |
| SiO <sub>2</sub>                                         | -ve                       | -ve                      |                             |                         |
| AgNPs                                                    | 16 ± 0.06*                | 21 ± 0.14*               |                             |                         |
| Ag/SiO <sub>2</sub> NC                                   | 14 ± 0.06*                | 24 ± 0.14*               |                             |                         |
| Fluconazole                                              | 15 ± 0*                   | 16 ± 0.14*               |                             |                         |

AgNPs and Penicillin G potassium showed a similar manner of MIC values (6.25 µg.mL<sup>-1</sup>) against *S. aureus*, *B. cereus*, and *E. coli* compared to Ag/SiO<sub>2</sub>NC (MIC value, 125 µg. mL<sup>-1</sup>). The

MIC values against *B. cinerea* were 6.25 and 25  $\mu\text{g.mL}^{-1}$  for Ag/SiO<sub>2</sub>NC and fluconazole, respectively. The better growth inhibition percentage of Ag/SiO<sub>2</sub>NC was against *B. cinerea* (50.2 %) followed by *B. cereus* (31.1 %), *S. aureus* (30.7 %), *E. coli* (26.6 %), and *C. albicans* (6.6 %) showing a dose-dependent manner of Ag/SiO<sub>2</sub>NC antimicrobial action (Fig. 7). The minimum antibacterial concentration of the Ag/SiO<sub>2</sub>NC is 0.2 and 0.3  $\mu\text{g.mL}^{-1}$  for *Bacillus* sp. and *E. coli*, respectively (Huang 2008). Qasim *et al.* (2015) suggested Ag/SiO<sub>2</sub>NC to be a potential antifungal

agent for *C. albicans* 077 showing that this tested human pathogenic fungus was sensitive to Ag/SiO<sub>2</sub>NC with MIC~6  $\mu\text{g.mL}^{-1}$  of Ag/SiO<sub>2</sub>NC. Vladkova *et al.* (2020) presented that TiO<sub>2</sub>/SiO<sub>2</sub>/Ag nanocomposite totally inhibited the *E. coli* growth within 30 min to 2 h. The growth of *B. cinerea* was almost completely inhibited (98.4 %) by Ag/SiO<sub>2</sub>NC (6.4  $\mu\text{g.mL}^{-1}$ ) treatment compared with AgNPs alone (72.43 %, 6.4  $\mu\text{g.mL}^{-1}$  as reported by Kim (2011).

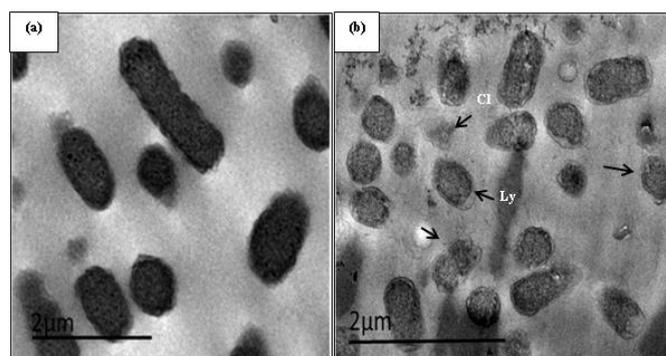


**Fig. 7.** Growth inhibition percentage of Ag/SiO<sub>2</sub>NC, AgNPs, and antibiotics at MIC values against *S. aureus*, *B. cereus*, *E. coli*, *C. albicans*, and *B. cinerea*.

It is known that both, Ag<sup>+</sup> and AgNPs are effective antimicrobial agents even though their antimicrobial mechanism is not fully understood (Kędziora *et al.* 2018). Several studies (Feng *et al.* 2000; Lara *et al.* 2011) reported the different mechanism of the antimicrobial action of nanomaterials such as penetrating the cell wall and plasma membrane, ending with damaging DNA molecules. Others suggested that nanomaterials might interact with thiol groups in proteins, which induces the inactivation of microbial proteins. In the presented study, AgNPs bonded on the surface of SiO<sub>2</sub> have an opposite charge with Gram-positive bacteria, in that way killing them more easily than Gram-negative bacteria due to the electrostatic attraction.

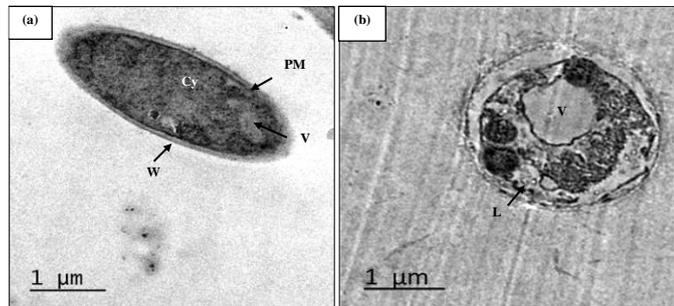
The antimicrobial activities of Ag/SiO<sub>2</sub>NC are investigated using *E. coli* and *B. cinerea* as two model microorganisms. As shown in Fig. 8, untreated *E. coli* was typically rod-shaped with smooth and intact cell walls. After being treated

with Ag/SiO<sub>2</sub>NC, cell walls became wrinkled and damaged. The separation between the bacterial cell wall and cell membrane was also noted.



**Fig. 8.** The antibacterial action of Ag/SiO<sub>2</sub>NC on the ultrastructure of *E. coli*. (a) A negative control (without Ag/SiO<sub>2</sub>NC). (b) A treated sample (150  $\mu\text{g.mL}^{-1}$ ), there are irregular rods (arrows) with lysed cell walls (Ly) and complete cell lysis (Cl). Also, note the separation that occurs between the bacterial cell wall and cell membrane.

With treated *B. cinerea* (Fig. 9), TEM micrographs showed many changes, including the reduced size of treated cells, the formation of a mucilage matrix connecting the hyphal cells together, the appearance of big vacuole and lipid droplets with severe leakage of cytoplasmic contents in comparing to the control. The separation between the fungal cell wall and plasma membrane was also detected in the treated cells. The observed damages of the *E. coli* and *B. cinerea* cells after the treatment by Ag/SiO<sub>2</sub>NC could be because of cellular interactions with the AgNPs. The combined action of adhesion and penetration of AgNPs might illustrate the biocidal action of the NC, plasma membrane being the target of rapid antimicrobial action of AgNPs in *E. coli* and *B. cinerea* (Rai *et al.* 2012). Eckhardt *et al.* (2013) reported that the binding of AgNPs with microbial proteins might inactivate the electron transport chain, in that way suppressing the respiration and growth of the microbial cells. To establish that the advantages of silver nanocomposites (AgNCs) and the possible mechanisms of their antimicrobial action outweigh the possible risks, the toxicity of AgNPs and AgNCs must be investigated.



**Fig. 9.** The antifungal activity of Ag/SiO<sub>2</sub>NC on the ultrastructure of *B. cinerea*. (a) negative control (without Ag/SiO<sub>2</sub>NC). Note normal cell wall (W), plasma membrane (PM), Vacuole (V), and compact cytoplasm (Cy). (b) The treated sample, note, the big vacuole (V) and lipid droplets (L). Also, note the separation that occurs between the fungal cell wall and plasma membrane (arrow).

## Conclusion

The embedded AgNPs in Ag/SiO<sub>2</sub>NC mediated by *E. coli* D8 were characterized as negative-charged (-31.0 mV) and spherical with an average size ranging between 18 and 24 nm. Ag/SiO<sub>2</sub>NC showed a good antimicrobial potential against Gram-negative and Gram-positive bacteria and

pathogenic yeast and fungi. The Ag/SiO<sub>2</sub>NC has brought many biomedical and agriculture applications (non-toxic to humans in minute concentrations). Further study will be designed to elucidate the mode of action of Ag/SiO<sub>2</sub>NC as an antimicrobial agent.

## Conflict of Interest

The authors declare that they have no conflict of interest.

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